The 2006 Water Quality Report

Drinking Water Quality

Since 1990, California water utilities have been providing an annual Water Quality Report to their customers. This year's report covers calendar year 2005 water quality testing, and has been prepared in compliance with regulations called for in the 1996 reauthorization of the Safe Drinking Water Act. The reauthorization charged the United States Environmental Protection Agency (USEPA) with updating and strengthening the tap water regulatory program and changed the report's due date to July 1.

USEPA and the California Department of Health Services (CDHS) are the agencies responsible for establishing drinking water quality standards. To ensure that your tap water is safe to drink, USEPA and CDHS prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. CDHS regulations also establish limits for contaminants in bottled water that must provide the same protection for public health. The federal Food and Drug Administration (FDA) also sets regulations for bottled water

The City of Newport Beach vigilantly safeguards its water supply and, as in years past, the water delivered to your home meets the standards required by the state and federal regulatory agencies. In some cases, your local utility goes beyond what is required to monitor for additional contaminants that have known health risks. For example, the Orange County Water District, which manages our groundwater basin, monitors our groundwater for the solvent 1,4-dioxane.

Unregulated contaminant monitoring helps USEPA determine where certain contaminants occur and whether it needs to establish regulations for those contaminants.

If you have any questions about your water, please contact us for answers...

For information about this report or your water quality information in general, please contact Pete Antista, Director of Utilities at (949) 718-3401. The City of Newport Beach Council meetings begin at 7:00 p.m. on the second and fourth Tuesday of each month and are open to the public. Meetings are held at the Council Chambers located at 3300 Newport Boulevard, Newport Beach. Matters from the public are heard at each meeting. Please feel free to participate in these meetings.

For more information about the health effects of the listed contaminants in the following tables, call the U.S. Environmental Protection Agency hotline at (800) 426-4791.

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Translate it, or speak with someone who understands it about your important information contains water.

Newport Beach Utilities Department 2006 Water Quality Report

The Continuing Quality of Your Water is Our Primary Concern

Immuno-Compromised People

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised people, such as those with cancer who are undergoing chemotherapy, persons who have had organ transplants, people with HIV/AIDS or other immune system disorders, some elderly persons and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care

Disinfection and Disinfection Byproducts

Disinfection of drinking water was one of the major public health advances in the 20th century. Disinfection was a major factor in reducing waterborne disease epidemics caused by pathogenic bacteria and viruses, and it remains an essential part of drinking water treatment today.

Chlorine disinfection has almost completely eliminated from our lives the risks of microbial waterborne diseases. Chlorine is added to your drinking water at the source of supply (groundwater well or surface water treatment plant). Enough chlorine is added so that it does not completely dissipate through the distribution system pipes. This "residual" chlorine helps to prevent the growth of bacteria in the pipes that carry drinking water from the source into your home.

However, chlorine can react with naturally-occurring materials in the water to form unintended chemical byproducts, called disinfection byproducts (DBPs), which may pose health risks. A major challenge is how to balance the risks from microbial pathogens and DBPs. It is important to provide protection from these microbial pathogens while simultaneously ensuring decreasing health risks from disinfection byproducts. The Safe Drinking Water Act requires the USEPA to develop rules to achieve these goals.

Trihalomethanes (THMs) and Haloacetic Acids (HAAs) are the most common and most studied DBPs found in drinking water treated with chlorine. In 1979, the USEPA set the maximum amount of total THMs allowed in drinking water at 100 parts per billion as an annual running average. Effective in January 2002, the Stage 1 Disinfectants / Disinfection Byproducts Rule lowered the total THM maximum annual average level to 80 parts per billion and added HAAs to the list of regulated chemicals in

Water Assessment Reports

Imported (Metropolitan) Water Assessment

In December 2002, Metropolitan Water District of Southern California completed its source water assessment of its Colorado River and State Water Project supplies. Colorado River supplies are considered to be most vulnerable to recreation, urban/storm water runoff, increasing urbanization in the watershed and wastewater. State Water Project supplies are considered to be most vulnerable to urban/storm water runoff, wildlife, agriculture, recreation and wastewater. A copy of the assessment can be obtained by contacting Metropolitan by phone at (213) 217-6850.

Groundwater Assessment

An assessment of the drinking water sources for City of Newport Beach Utilities Department was completed in December 2002. The groundwater sources are considered most vulnerable to the following activities not associated with detected contaminants: Dry cleaners, gas stations, and known contaminant plumes.

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Want Additional Information?

There's a wealth of information on the internet about Drinking Water Quality and water issues in general. Some good sites both local and national — to begin your own investigation are:

> Municipal Water District of Orange County www.mwdoc.com

> > Orange County Water District

www.ocwd.com

Metropolitan Water District of Southern California www.mwdh2o.com

California Department of Health Services, Division of Drinking

Water and Environmental Management www.dhs.ca.gov/ps/ddwem

ILS Environmental Protection Agency www.epa.gov/safewater/

Table Definitions

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs (2nd MCL) are set to protect the odor, taste, and appearance of drinking water.

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency.

MRDL (Maximum Residual Disinfectant Level): The level of a disinfectant added for water treatment that may not be exceeded at a consumer's tap.

MRDLG (Maximum Residual Disinfectant Level Goal): The level of a disinfectant added for water treatment below which there is no known or expected risk to health. MRDLGs are set by the USEPA.

PHG (Public Health Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

Primary Drinking Water Standard or PDWS: MCLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

TT (Treatment Technique): A required process intended to reduce the level of a contaminant in drinking water.

Regulatory Action Level: The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Measurements: Water is sampled and tested throughout the year. Contaminants are measured in parts per million (ppm), parts per billion (ppb), parts per trillion (ppt), and even parts per quadrillion (ppq). If this is difficult to imagine, think about these comparisons: Parts per million (mg/L):

1 second in 12 days

Parts per billion (µg/L):

- 1 second in 32 years
- 1 penny in \$10,000 • 1 inch in 16 miles
- 1 penny in \$10 million • 1 inch in 16,000 miles

It is important to note, however, that even a small concentration of certain contaminants can adversely affect a water supply.

The State allows us to monitor for some contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of our data, though representative, are more than one year old.

drinking water. Your drinking water complies with the Stage 1 Disinfectants / Disinfection Byproducts Rule. In 2003, the USEPA proposed a Stage 2regulation that will further control allowable levels of DBPs in drinking water without compromising disinfection itself. This regulation was finalized by USEPA in January 2006.

Infants and young children are typically more vulnerable to lead in drinking water than the general population. It is possible that lead levels at your home may be higher than at other homes in the community as a result of materials used in your home's plumbing. If you are concerned about elevated lead levels in your home's water, you may wish to have your water

tested and flush your tap for 30 seconds to 2 minutes before using tap water. Additional information is available from the Safe Drinking Water Hotline (800) 426-4791.

Arsenic

Arsenic is an element that occurs in the earth's crust. Accordingly, there are natural sources of exposure. Exposure to arsenic at high levels can pose serious health effects, as it is known to cause skin cancer and other cancers of the internal organs. In addition, it has been reported to affect the vascular

*Contaminant is regulated by a secondary standard

system and has been associated with the development of diabetes. The USEPA established a maximum contaminant level (MCL) for arsenic of 50 parts per billion in 1975. In January 2002, USEPA adopted a new standard for arsenic in drinking water that requires public water supplies to reduce arsenic to 10 parts per billion by January 2006. Groundwater and imported water

supplies in Orange County generally range between non-detectable levels and 5 parts per billion.

The California Health & Safety Code requires the CDHS to adopt a new California MCL for arsenic by June 30, 2004. Although this deadline has passed without a proposed regulation, it is anticipated that the CDHS could propose a California arsenic MCL which is lower than the federal MCL sometime in 2006.

Nitrate

The maximum allowable level of nitrate in drinking water, also called the maximum contaminant level or MCL, is 45 milligrams per liter as nitrate

> (mg/L as NO_3). The nitrate MCL can also be expressed as 10 milligrams per liter as nitrogen (mg/L as N). Both numbers are equivalent values. At times, nitrate in your tap water may have exceeded one-half the MCL, but it was never greater than the MCL. The following advisory is issued because in 2005 we recorded nitrate measurements in the drinking water supply which exceeded onehalf the nitrate MCL.

> Nitrate in drinking water at levels above 45 mg/L (or the equivalent 10 mg/L as N) is a health risk for infants of less than six months of age. Such nitrate levels in drinking water can interfere with the capacity of the infant's blood to carry oxygen, resulting in a serious illness; symptoms

include shortness of breath and blueness of the skin. Nitrate levels above 45 parts-per-million may also affect the ability of the blood to carry oxygen in other individuals, such as pregnant women and those with certain specific enzyme deficiencies. If you are caring for an infant, or you are pregnant, you should ask advice from your health care provider.

2005 City of Newport Beach Groundwater Quality

2003 City of Newport Beach Groundwater Quanty								
Chemical	MCL	PHG (MCLG)	Average Amount	Range of Detections	MCL Violation?	Most Recent Sampling Date	Typical Source of Contaminant	
Radiologicals								
Alpha Radiation (pCi/L)	15	(0)	7.1	ND - 16	No	2005	Erosion of Natural Deposits	
Uranium (pCi/L)	20	0.43	7.3	2.1 – 15	No	2005	Erosion of Natural Deposits	
norganic Chemicals								
Barium (ppm)	1	2	<0.1	ND - 0.11	No	2005	Erosion of Natural Deposits	
Fluoride (ppm)	2	1	0.32	0.29 - 0.37	No	2005	Erosion of Natural Deposits	
Nitrate (ppm as NO ₃)	45	45	9.2	ND - 25	No	2005	Fertilizers, Septic Tanks	
Nitrate+Nitrite (ppm as N)	10	10	2.1	ND - 5.7	No	2005	Fertilizers, Septic Tanks	
Secondary Standards*								
Chloride (ppm)	500*	n/a	58	32 – 90	No	2005	Erosion of Natural Deposits	
Color (color units)	15*	n/a	3	3	No	2005	Erosion of Natural Deposits	
Specific Conductance (µmho/cm)	1,600*	n/a	766	468 - 1,140	No	2005	Erosion of Natural Deposits	
Sulfate (ppm)	500*	n/a	113	39 – 212	No	2005	Erosion of Natural Deposits	
Total Dissolved Solids (ppm)	1000*	n/a	475	242 – 752	No	2005	Erosion of Natural Deposits	
Turbidity (ntu)	5*	n/a	0.6	0.4 - 1.0	No	2005	Erosion of Natural Deposits	
Jnregulated Contaminants R	Requiring Monitor	ing						
Alkalinity (ppm as CaCO ₃)	Not Regulated	n/a	178	133 – 224	n/a	2005	Erosion of Natural Deposits	
Bicarbonate (ppm)	Not Regulated	n/a	217	159 – 273	n/a	2005	Erosion of Natural Deposits	
Boron (ppb)	Not Regulated	n/a	110	100 - 130	n/a	2005	Erosion of Natural Deposits	
Calcium (ppm)	Not Regulated	n/a	85	31 – 145	n/a	2005	Erosion of Natural Deposits	
Hardness (ppm as CaCO ₃)	Not Regulated	n/a	271	91 – 466	n/a	2005	Erosion of Natural Deposits	
Magnesium (ppm)	Not Regulated	n/a	14	3.6 - 25	n/a	2005	Erosion of Natural Deposits	
pH (units)	Not Regulated	n/a	8.2	8.0 - 8.3	n/a	2005	Acidity,hydrogen ions	
Potassium (ppm)	Not Regulated	n/a	3.2	2.0 - 4.6	n/a	2005	Erosion of Natural Deposits	
Sodium (ppm)	Not Regulated	n/a	57	49 – 61	n/a	2005	Erosion of Natural Deposits	
/anadium (ppb)	Not Regulated	n/a	<3	ND - 3.7	n/a	2005	Erosion of Natural Deposits	

ppb = parts-per-billion; ppm = parts-per-million; pCi/L = picoCuries per liter; ntu = nephelometric turbidity units; ND = not detected; n/a = not applicable; < = average is less than the detection limit for reporting purposes; MCL = Maximum Contaminant Level; (MCLG) = federal MCL Goal; PHG = California Public Health Goal µmho/cm = micromho per centimeter; *Contaminant is regulated by a secondary standard to maintain aesthetic qualities (taste, odor, color)

2005 City of Newport Beach Distribution System Water Quality

Disinfection Byproducts	MCL (MRDL/MRDLG)	Average Amount	Range of Detections	MCL Violation?	Typical Source of Contaminant
Total Trihalomethanes (ppb)	80	50	5 – 108	No	Byproducts of chlorine disinfection
Haloacetic Acids (ppb)	60	20	ND - 44	No	Byproducts of chlorine disinfection
Chlorine Residual (ppm)	(4 / 4)	1.5	ND - 2.8	No	Disinfectant added for treatment
Aesthetic Quality					
Odor (threshold odor number)	3*	1	1	No	Erosion of Natural Deposits
Turbidity (ntu)	5*	0.2	0.1 – 1.0	No	Erosion of Natural Deposits

Twelve locations in the distribution system are tested quarterly for total trihalomethanes and haloacetic acids; thirty locations are tested monthly for color, odor and turbidity; color was not detected in 2005. MRDL = Maximum Residual Disinfectant Level; MRDLG = Maximum Residual Disinfectant Level; MRDLG = Maximum Residual Disinfectant Level;

Lead and Copper Action Levels at Residential Taps

	Action Level (AL)	Health Goal	90th Percentile Value	Sites Exceeding AL / Number of Sites	AL Violation?	Typical Source of Contaminant
Copper (ppm)	1.3	0.17	0.3	0 / 30	No	Corrosion of household plumbing
Lead (ppb)	15	2	9	2 / 30	No	Corrosion of household plumbing

Lead was detected in five homes; two exceeded the regulatory action level. Copper was detected in 26 samples; none exceeded the action level.

A regulatory action level is the concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow

2005 Metropolitan Water District of Southern California Treated Surface Water

	PHG, or	Average	Range of	MCL		
Chemical	MCL	(MCLG)	Amount	Detections	Violation?	Typical Source of Contaminant
Radiologicals – Tested in 2005	5					
Alpha Radiation (pCi/L)	15	(0)	<3	ND - 3.2	No	Decay of man-made or natural deposits
Beta Radiation (pCi/L)	50	(0)	4.8	ND - 6.4	No	Erosion of natural deposits
Inorganic Chemicals – Tested	in 2005					
Aluminum (ppm)	1 / 0.2*	0.6	< 0.05	ND - 0.1	No	Erosion of natural deposits
Barium (ppm)	1	2	<0.1	ND - 0.1	No	Erosion of natural deposits
Fluoride (ppm)	2	1	0.19	0.15 - 0.22	No	Erosion of natural deposits
Nitrate as NO ₃ (ppm)	45	45	2.3	ND - 3.6	No	Agriculture runoff and sewage
Nitrate and Nitrite as N (ppm)	10	10	0.5	ND - 0.8	No	Agriculture runoff and sewage
Secondary Standards* – Teste	ed in 2005					
Chloride (ppm)	500*	n/a	77	67 – 85	No	Runoff or leaching from natural deposits
Color (color units)	15*	n/a	2	1 – 2	No	Runoff or leaching from natural deposits
Corrosivity (LSI)	non-corrosive	n/a	0.27	0.15 - 0.39	No	Elemental balance in water
Odor (odor units)	3*	n/a	2	2	No	Naturally-occurring organic materials
Specific Conductance (µmho/cm)	1,600*	n/a	792	734 – 871	No	Substances that form ions in water
Sulfate (ppm)	500*	n/a	171	151- 202	No	Runoff or leaching of natural deposits
Total Dissolved Solids (ppm)	1,000*	n/a	468	426 - 528	No	Runoff or leaching of natural deposits
Turbidity (NTU)	5*	n/a	0.06	0.05 - 0.07	No	Runoff or leaching of natural deposits
Unregulated Chemicals – Test	ted in 2005					
Alkalinity (ppm)	Not Regulated	n/a	91	83 – 101	n/a	Runoff or leaching from natural deposits
Boron (ppb)	Not Regulated	n/a	160	130 - 200	n/a	Runoff or leaching from natural deposits
Calcium (ppm)	Not Regulated	n/a	45	39 – 53	n/a	Runoff or leaching from natural deposits
Hardness, total (ppm)	Not Regulated	n/a	197	176 – 225	n/a	Runoff or leaching of natural deposits
Hardness, total (grains/gal)	Not Regulated	n/a	12	10 – 13	n/a	Runoff or leaching of natural deposits
Magnesium (ppm)	Not Regulated	n/a	21	19 – 23	n/a	Runoff or leaching from natural deposits
N-Nitrosodimethylamine (ppt)	Not Regulated	n/a	<2	ND - 2.2	n/a	By-product of drinking water chlorination
pH (pH units)	Not Regulated	n/a	8.2	8.1 – 8.2	n/a	Hydrogen ion concentration
Potassium (ppm)	Not Regulated	n/a	3.8	3.5 – 4.1	n/a	Runoff or leaching from natural deposits
Sodium (ppm)	Not Regulated	n/a	82	73 – 90	n/a	Runoff or leaching from natural deposits
Vanadium (ppb)	Not Regulated	n/a	3.3	3.2 - 3.4	n/a	Runoff or leaching from natural deposits

ppb = parts-per-billion; ppm = parts-per-million; ppt = parts-per-trillion; pCi/L = picoCuries per liter; ntu = nephelometric turbidity units; µmho/cm = micro ND = not detected; < = average is less than the detection limit for reporting purposes; MCL = Maximum Contaminant Level; (MCLG) = federal MCL Goal; PHG = California Public Health Goal; n/a = not applicable; LSI = Langelier Saturation Index; *Contaminant is regulated by a secondary standard.

Treatment Technique Turbidity Measurements **Typical Source of Contaminant** 1) Highest single turbidity measurement 1 NTU 0.06 Soil run-off Soil run-off 2) Percentage of samples less than 0.3 NTU 95% 100% No Turbidity is a measure of the cloudiness of the water, an indication of particulate matter, some of which might include harmful microorganisms.

Low turbidity in Metropolitan's treated water is a good indicator of effective filtration. Filtration is called a "treatment technique."

A treatment technique is a required process intended to reduce the level of contaminants in drinking water that are difficult and sometimes impossible to measure directly.

What You Need to Know About Your Water, and How it May Affect You

Sources of Supply

range County's water supplies are a blend of groundwater provided by the Orange County Water District (OCWD) and water imported from Northern California and the Colorado River by the Municipal Water District of Orange County (MWDOC) via the Metropolitan Water District of Southern California (MET). Groundwater comes from a natural underground aguifer that is replenished with water from the Santa Ana River, local rainfall and imported water. The groundwater basin is 350 square miles and lies beneath north and central Orange County from Irvine to the Los Angeles border and from Yorba Linda to the Pacific Ocean. More than 20 cities and retail water districts draw from the basin to provide water to homes and businesses. In south Orange County, nearly 100 percent of the water is imported and delivered to the cities and retail water districts, where it is stored in above-ground reservoirs and tanks before being sent to homes and businesses.

Orange County's Water Future

For years, Orange County has enjoyed an abundant, seemingly endless supply of high-quality water. However, as water demand continues to increase statewide, we must be even more conscientious about our water supply and maximize the efficient use of this precious natural resource.

OCWD and MWDOC work cooperatively to evaluate new and innovative water management and supply development programs, including water reuse and recycling, wetlands expansion, recharge facility construction, ocean and brackish water desalination, surface storage and water use efficiency programs. These efforts are

helping to enhance long-term countywide water reliability and water quality.

A healthy water future for Orange County rests on finding and developing new water supplies, as well as protecting and improving the quality of the water that we have today.

Your local and regional water agencies are committed to making the necessary investments today in new water management projects to ensure an abundant and high-quality water supply for our future.

Engineering
marvels, the State
Water Project and
Colorado River Aqueduct,
make our way of life possible Angeles
by delivering water to millions
of people in Orange County.

Our future.

State Water Project

LA. Aqueduct

Colorado River
Aqueduct

Big Bear Reservoir

San Bernardino

Cascading from its source high in the San Bernardino

Mountains, the Santa Ana River is fed by a watershed over 2,500 square miles in area. River water is captured behind Prado Dam and slowly

Seven released to help replenish the Orange County Qaks groundwater basin. Percolation ponds in Anaheim and Orange hold this water so it can seep into the basin.

Prado Dam

Anaheim and Orange hold this water so it can seep into the basin.

Basic Information About Drinking Water Contaminants

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs and wells. As water travels over the surface of land or through the layers of the ground it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animal and human activity.

Contaminants that may be present in source water include:

- Microbial contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations and wildlife.
- ▶ Inorganic contaminants, such as salts and metals, which can be naturally occurring or result from urban storm runoff, industrial or domestic wastewater discharges, oil and gas production, mining and farming.
- Radioactive contaminants, which can be naturally occurring or be the result of oil and gas production or mining activities.
- Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gasoline stations, urban storm water runoff, agricultural application and septic systems.

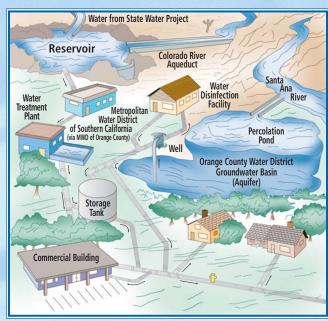
In order to ensure that tap water is safe to drink, USEPA and the CDHS prescribe regulations that limit the amount of certain

contaminants in water provided by public water systems. CDHS regulations also establish limits for contaminants in bottled water that must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline at (800) 426-4791.

Cryptosporidium

Cryptosporidium is a microscopic organism that, when ingested, can cause diarrhea, fever, and other gastrointestinal symptoms. The organism comes from animal and/or human wastes and may be in surface water. The Metropolitan Water District of Southern California tested their treated water for Cryptosporidium in 2005 but did not detect it. Any Cryptosporidium in Metropolitan's source water is eliminated by an effective treatment combination including sedimentation, filtration and disinfection.

The USEPA and the federal Centers for Disease Control guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from USEPA's Safe Drinking Water Hotline at (800) 426-4791 between 9 a.m. and 5 p.m. Eastern Time (6 a.m. to 2 p.m. in California).



Imported water — from the Colorado River and northern California — travels hundreds of miles to meet the needs of Orange County. Water is also pumped from the groundwater basin that spans 350 square miles under north and central Orange County.

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(mg/L as NO₃). The nitrate MCL can also be expressed as 10 milligrams per liter as nitrogen (mg/L as N). Both numbers are equivalent values. At times, nitrate in your tap water may have exceeded one-half the MCL, but it was never greater than the MCL. The following advisory is issued because in 2005 we recorded nitrate measurements in the drinking water supply which exceeded one-half the nitrate MCL.

Nitrate in drinking water at levels above 45 mg/L (or the equivalent 10 mg/L as N) is a health risk for infants of less than six months of age. Such nitrate levels in drinking water can interfere with the capacity of the infant's blood to carry oxygen, resulting in a serious illness; symptoms

include shortness of breath and blueness of the skin. Nitrate levels above 45 parts-per-million may also affect the ability of the blood to carry oxygen in other individuals, such as pregnant women and those with certain specific enzyme deficiencies. If you are caring for an infant, or you are pregnant, you should ask advice from your health care provider.

2005 City of Newport Beach Groundwater Quality

Chemical	MCL	PHG (MCLG)	Average Amount	Range of Detections	MCL Violation?	Most Recent Sampling Date	Typical Source of Contaminant
Radiologicals		(/				, J	
Alpha Radiation (pCi/L)	15	(0)	7.1	ND - 16	No	2005	Erosion of Natural Deposits
Uranium (pCi/L)	20	0.43	7.3	2.1 – 15	No	2005	Erosion of Natural Deposits
Inorganic Chemicals							
Barium (ppm)	1	2	<0.1	ND - 0.11	No	2005	Erosion of Natural Deposits
Fluoride (ppm)	2	1	0.32	0.29 - 0.37	No	2005	Erosion of Natural Deposits
Nitrate (ppm as NO ₃)	45	45	9.2	ND - 25	No	2005	Fertilizers, Septic Tanks
Nitrate+Nitrite (ppm as N)	10	10	2.1	ND - 5.7	No	2005	Fertilizers, Septic Tanks
Secondary Standards*							
Chloride (ppm)	500*	n/a	58	32 – 90	No	2005	Erosion of Natural Deposits
Color (color units)	15*	n/a	3	3	No	2005	Erosion of Natural Deposits
Specific Conductance (µmho/cm)	1,600*	n/a	766	468 - 1,140	No	2005	Erosion of Natural Deposits
Sulfate (ppm)	500*	n/a	113	39 – 212	No	2005	Erosion of Natural Deposits
Total Dissolved Solids (ppm)	1000*	n/a	475	242 – 752	No	2005	Erosion of Natural Deposits
Turbidity (ntu)	5*	n/a	0.6	0.4 - 1.0	No	2005	Erosion of Natural Deposits
Unregulated Contaminants I	Requiring Monitor	ing					
Alkalinity (ppm as CaCO ₃)	Not Regulated	n/a	178	133 – 224	n/a	2005	Erosion of Natural Deposits
Bicarbonate (ppm)	Not Regulated	n/a	217	159 – 273	n/a	2005	Erosion of Natural Deposits
Boron (ppb)	Not Regulated	n/a	110	100 – 130	n/a	2005	Erosion of Natural Deposits
Calcium (ppm)	Not Regulated	n/a	85	31 – 145	n/a	2005	Erosion of Natural Deposits
Hardness (ppm as CaCO ₃)	Not Regulated	n/a	271	91 – 466	n/a	2005	Erosion of Natural Deposits
Magnesium (ppm)	Not Regulated	n/a	14	3.6 - 25	n/a	2005	Erosion of Natural Deposits
pH (units)	Not Regulated	n/a	8.2	8.0 - 8.3	n/a	2005	Acidity,hydrogen ions
Potassium (ppm)	Not Regulated	n/a	3.2	2.0 - 4.6	n/a	2005	Erosion of Natural Deposits
Sodium (ppm)	Not Regulated	n/a	57	49 – 61	n/a	2005	Erosion of Natural Deposits
Vanadium (ppb)	Not Regulated	n/a	<3	ND - 3.7	n/a	2005	Erosion of Natural Deposits

ppb = parts-per-billion; ppm = parts-per-million; pCi/L = picoCuries per liter; ntu = nephelometric turbidity units; ND = not detected; n/a = not applicable; <= average is less than the detection limit for reporting purposes; MCL = Maximum Contaminant Level; (MCLG) = federal MCL Goal; PHG = California Public Health Goal umb/o/cm = micromho per centimeter; "Contaminant is regulated by a secondary standard to maintain aesthetic qualities (taste, odor, color).

2005 City of Newport Beach Distribution System Water Quality

Disinfection	MCL	Average	Range of	MCL	Typical Source
Byproducts	(MRDL/MRDLG)	Amount	Detections	Violation?	of Contaminant